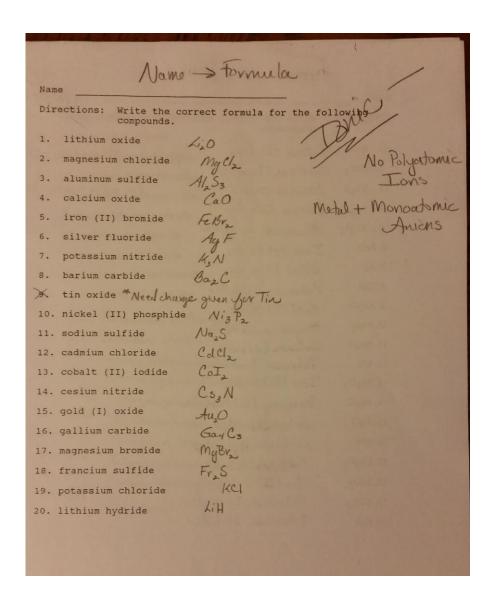
## **Ionic Compounds**

Formula -> Name (Two Practice Pages)



Directions:	Write the correct name for the following compounds.  Sadium Sulfate
1. Na <sub>2</sub> SO <sub>4</sub>	Salium Sulfate
2. Ca(NO <sub>3</sub> ) <sub>2</sub>	: Calcium Nitrate
3. $(NH_4)_2 S$	Ammonium Sulfide Metal + Polyadinue I
4. Cu <sub>2</sub> co <sub>3</sub>	Copper (I) Carbonate
5. Cu(NO <sub>2</sub> ) <sub>2</sub>	Copper (II) Nitrite
6. Li <sub>3</sub> PO <sub>4</sub>	Lithium Phosphate
7. KC103	Potassium Chlorate
H <sub>2</sub> O <sub>2</sub>	Hydrogen Peroxide 02 = Peroxide
9. MgsO <sub>3</sub>	Magnesium Sulfite
10. Ba(OH) <sub>2</sub>	Barium Hydroxide
11. znso <sub>3</sub>	Zinc Sulfite
12. Fe <sub>2</sub> (CrO <sub>4</sub> ) <sub>3</sub>	3 Iron (III) Chromate
13. Frc <sub>2</sub> H <sub>3</sub> O <sub>2</sub>	Francium Acetale
14. Ca(MnO <sub>4</sub> ) <sub>2</sub>	Calcium Permanganate
15. CoIO3	Cobalt (I) Dodate